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Section : Physics

A uniformly tapering conical wire is made from a material of Young's modulus *Y* and has a normal, unextended length *L*. The radii, at the upper and lower ends of this conical wire, have values *R* and 3*R*, respectively. The upper end of the wire is fixed to a rigid support and a mass *M* is suspended from its lower end. The equilibrium extended length, of this wire, would equal:

Chosen Option : --

Options

$$L\left(1+\frac{2}{9}\,\frac{Mg}{\pi YR^2}\right)$$

$$2. L \left(1 + \frac{1}{9} \frac{Mg}{\pi YR^2}\right)$$

3.
$$L\left(1+\frac{2}{3}\frac{Mg}{\pi YR^2}\right)$$

4
$$L\left(1+\frac{1}{3}\frac{Mg}{\pi YR^2}\right)$$

Q.2 An unknown transistor needs to be identified as a *npn* or *pnp* type. A multimeter, with + ve and - ve terminals, is used to measure resistance between different terminals of transistor. If terminal 2 is the base of the transistor then which of the following is correct for a *pnp* transistor?

+ ve terminal 2, – ve terminal 1, resistance high

2. + ve terminal 3, -ve terminal 2, resistance high

Chosen Option : 4

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- +ve terminal 2, -ve terminal
- 3, resistance low
- + ve terminal 1, -ve terminal
 - 2, resistance high
- Q.3 In the following 'I' refers to current and other symbols have their usual meaning. Choose the option that corresponds to the dimensions of electrical conductivity:

Chosen Option: 3

Optio ns 1 ML - 3 T - 3 I²

- $^{2} M^{-1} L^{3} T^{3} I$
- $^{3} M^{-1} L^{-3} T^{3} I^{2}$
- $^{4}~{
 m M}^{-1}\,{
 m L}^{-3}\,{
 m T}^{3}\,{
 m I}$
- Q.4 A hydrogen atom makes a transition from n=2 to n=1 and emits a photon. This photon strikes a doubly ionized lithium atom (z=3) in excited state and completely removes the orbiting electron. The least quantum number for the excited state of the ion for the process is:

Chosen Option: 2

Optio 1 5

- 2. 4
- 3. 2
- 4. 3
- Q.5 Three capacitors each of 4 μ F are to be connected in such a way that the effective capacitance is 6 μ F. This can be done by connecting them:

Chosen Option: 4

Optio 1 all in parallel

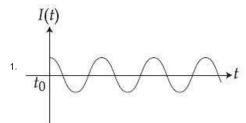
- 2 two in parallel and one in series
- all in series
- 4 two in series and one in parallel
- Q.6 A series LR circuit is connected to a voltage source with $V(t) = V_0 \sin \Omega t$. After very large time, current I(t) behaves as $\left(t_0 >> \frac{L}{R}\right)$:

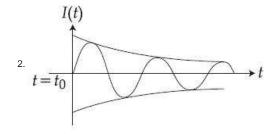
Chosen Option : --

Optio ns

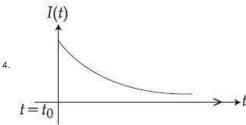
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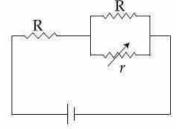
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Chosen Option: 1

In the circuit shown, the resistance r is a variable resistance. If for r=fR, the heat generation in r is maximum then the value of f is:

3.
$$\frac{3}{4}$$

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- $4. \frac{1}{2}$
- A car of weight W is on an inclined road that rises by 100 m over a distance of 1 km and applies a constant frictional force $\frac{W}{20}$

Chosen Option : --

on the car. While moving uphill on the road at a speed of 10 ms^{-1} , the car needs power

P. If it needs power $\frac{P}{2}$ while moving downhill at speed v then value of v is :

- Optio 1. 15 ms⁻¹
 - 2. 10 ms⁻¹
 - 3. 20 ms⁻¹
 - $^{4} 5 \, \text{ms}^{-1}$
- Q.9 A magnetic dipole is acted upon by two magnetic fields which are inclined to each other at an angle of 75°. One of the fields has a magnitude of 15 mT. The dipole attains stable equilibrium at an angle of 30° with this field. The magnitude of the other field (in mT) is close to:

Chosen Option : 4

Optio 1 1060

- 2 36
- 3. 1
- 4. 11
- Two particles are performing simple harmonic motion in a straight line about the same equilibrium point. The amplitude and time period for both particles are same and equal to *A* and *T*, respectively. At time *t* = 0 one particle has displacement *A* while

the other one has displacement $\frac{-A}{2}$ and

they are moving towards each other. If they cross each other at time t, then t is:

Optio T

Chosen Option : --



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2. $\frac{T}{3}$ 3. $\frac{T}{6}$ 4. $\frac{5T}{6}$	
An experiment is performed to determine the I - V characteristics of a Zener diode, which has a protective resistance of R = $100~\Omega$, and a maximum power of dissipation rating of $1~W$. The minimum voltage range of the DC source in the circuit is:	Chosen Option :
$_{\rm ns}^{\rm Optio}$ 1. $0-5~{ m V}$	
2. 0 – 24 V	
3. 0 – 8 V	
4. 0 – 12 V	
$A 50 \Omega$ resistance is connected to a battery of $5 \ V$. A galvanometer of resistance $100 \ \Omega$ is to be used as an ammeter to measure current through the resistance, for this a resistance r_s is connected to the galvanometer. Which of the following connections should be employed if the measured current is within 1% of the current without the ammeter in the circuit?	Chosen Option :
$r_s = 1 \Omega$ in series with galvanometer	
$r_s = 0.5 \Omega$ in parallel with the galvanometer	
3. $r_s = 1 \Omega$ in parallel with galvanometer	
$r_s = 0.5 \Omega$ in series with the galvanometer	
Q.13 200 g water is heated from 40°C to 60°C. Ignoring the slight expansion of water, the change in its internal energy is close to (Given specific heat of	Chosen Option : 4



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- 2. 4.2 kJ
- 3. 8.4 kJ
- 4 16.7 kJ

O.14 In Young's double slit experiment, the distance between slits and the screen is 1.0 m and monochromatic light of 600 nm is being used. A person standing near the slits is looking at the fringe pattern. When the separation between the slits is varied, the interference pattern disappears for a particular distance d_0 between the slits. If

the angular resolution of the eye is $\frac{1}{60}$,

the value of d_0 is close to :

Optio ns 1. 4 mm

- 2. 1 mm
- 3. 3 mm
- 4. 2 mm

Q.15 To know the resistance G of a galvanometer by half deflection method, a battery of emf V_F and resistance R is used to deflect the galvanometer by angle θ . If a shunt of resistance S is needed to get half deflection then G, R and S are related by the equation:

Optio 1 2G = S

- 2. 2S(R+G) = RG
- 3. 2S = G
- 4. S(R+G) = RG

Q.16

Chosen Option: 1

Chosen Option: 4

Chosen Option: --



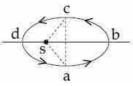
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Figure shows elliptical path abcd of a planet around the sun S such that the area of

triangle csa is $\frac{1}{4}$ the area of the ellipse. (See

figure) With db as the semimajor axis, and ca as the semiminor axis. If t_1 is the time taken for planet to go over path abc and t_2 for path taken over cda then:



Optio 1. $t_1 = 3t_2$

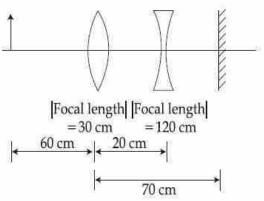
2. $t_1 = 4t_2$

3. $t_1 = t_2$

4. $t_1 = 2t_2$

Q.17

A convex lens, of focal length 30 cm, a concave lens of focal length 120 cm, and a plane mirror are arranged as shown. For an object kept at a distance of 60 cm from the convex lens, the final image, formed by the combination, is a real image, at a distance of:



Optio 1. 60 cm from the concave lens

- 2 70 cm from the concave lens
- 3. 70 cm from the convex lens
- 4.60 cm from the convex lens

Q.18

Chosen Option: --

Chosen Option: 4



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Two engines pass each other moving in opposite directions with uniform speed of 30 m/s. One of them is blowing a whistle of frequency 540 Hz. Calculate the frequency heard by driver of second engine before they pass each other. Speed of sound is 330 m/sec:

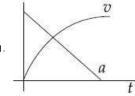
Optio 1 270 Hz

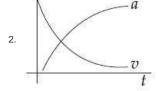
- 2 540 Hz
- 3. 450 Hz
- 4 648 Hz

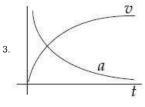
Which of the following option correctly describes the variation of the speed v and acceleration 'a' of a point mass falling vertically in a viscous medium that applies a force F = -kv, where 'k' is a constant, on the body ? (Graphs are schematic and not drawn to scale)

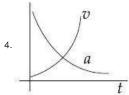
Chosen Option : --

Optio









Q.20

Chosen Option : 2



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A B Y 0 0 0 0 1 1 1 0 1 1 1 1 1 1 NAND - Gate 2 OR - Gate 3 NOR - Gate 4 AND - Gate 4 And - Gate The mass m and a metallic wire of negligible mass has time period 2 s at T = 0°C. If the	
0 1 1 1 0 1 1 1 1 1 1 1 NAND - Gate OR - Gate NOR - Gate AND - Gate A simple pendulum made of a bob of mass m and a metallic wire of negligible	
1 0 1 1 1 1 NAND - Gate OR - Gate NOR - Gate AND - Gate A simple pendulum made of a bob of mass m and a metallic wire of negligible	
1 1 1 tio 1 NAND - Gate 2 OR - Gate 3 NOR - Gate 4 AND - Gate A simple pendulum made of a bob of mass m and a metallic wire of negligible Chosen Option:	
2. OR - Gate 3. NOR - Gate 4. AND - Gate A simple pendulum made of a bob of mass m and a metallic wire of negligible	
2. OR - Gate 3. NOR - Gate 4. AND - Gate A simple pendulum made of a bob of mass m and a metallic wire of negligible Chosen Option:	
3. NOR - Gate 4. AND - Gate 21 A simple pendulum made of a bob of mass m and a metallic wire of negligible Chosen Option:	
4. AND - Gate 21 A simple pendulum made of a bob of mass m and a metallic wire of negligible Chosen Option:	
4. AND - Gate 21 A simple pendulum made of a bob of mass m and a metallic wire of negligible Chosen Option:	
A simple pendulum made of a bob of Chosen Option: mass m and a metallic wire of negligible	
A simple pendulum made of a bob of Chosen Option:	
THOSE THIS CHIEF PETROCE I TO CO. II CHE	
temperature of the wire is increased and the	
corresponding change in its time period is	
plotted against its temperature, the	
resulting graph is a line of slope S. If the	
coefficient of linear expansion of metal is α	
then the value of S is :	
$_{ m s}^{ m tio}$ 1. $2lpha$	
The control	
$2.\frac{\alpha}{2}$	
3. 1	
3. - α	
4. α	
An audio signal consists of two distinct Chosen Option:4	
sounds : one a human speech signal in the	
frequency band of 200 Hz to 2700 Hz, while	
the other is a high frequency music signal	
in the frequency band of 10200 Hz to	
15200 Hz. The ratio of the AM signal	
bandwidth required to send both the	
signals together to the AM signal	
bandwidth required to send just the human	
speech is:	
tio 1. 2	
2. 53. 3	



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To find the focal length of a convex mirror,
a student records the following data:

Object Pin	Convex Lens	Convex Mirror	Image Pin
22.2 cm	32.2 cm	45.8 cm	71.2 cm

The focal length of the convex lens is f_1 and that of mirror is f_2 . Then taking index correction to be negligibly small, f_1 and f_2 are close to:

Optio s $f_1 = 7.8 \text{ cm}$

$$f_2 = 25.4 \text{ cm}$$

 $f_1 = 12.7 \text{ cm}$ $f_2 = 7.8 \text{ cm}$

$$f_2 = 7.8 \text{ cm}$$

3. $f_1 = 7.8 \text{ cm}$ $f_2 = 12.7 \text{ cm}$

$$f_2 = 12.7 \text{ cm}$$

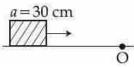
4. $f_1 = 15.6 \text{ cm}$ $f_2 = 25.4 \text{ cm}$

$$f_2 = 25.4 \text{ cm}$$

Q.24 A cubical block of side 30 cm is moving with velocity 2 ms⁻¹ on a smooth horizontal surface. The surface has a bump at a point O as shown in figure. The angular velocity (in rad/s) of the block immediately after it hits the bump, is:

Chosen Option: 3

Chosen Option: --



Optio 1 5.0

2. 6.7

3. 13.3

4. 9.4

Q.25 A rocket is fired vertically from the earth with an acceleration of 2g, where g is the gravitational acceleration. On an inclined plane inside the rocket, making an angle θ with the horizontal, a point object of mass m is kept. The minimum coefficient of friction μ_{\min} between the mass and the inclined surface such that the mass does not move is:

Chosen Option: 3

Optio 1 $\tan \theta$

2 $tan2\theta$

3. $2 \tan \theta$

4 3 $\tan \theta$

Q.26

Chosen Option : -



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The potential (in volts) of a charge distribution is given by

$$V(z) = 30 - 5z^2$$
 for $|z| \le 1$ m

$$V(z) = 35 - 10 |z| \text{ for } |z| \ge 1 \text{ m.}$$

V(z) does not depend on x and y. If this potential is generated by a constant charge per unit volume ρ_0 (in units of ϵ_0) which is spread over a certain region, then choose the correct statement.

optio 1 $\rho_0 = 40 \epsilon_0$ in the entire region

- $_{2}$ $\rho_{0}\!=\!10$ ϵ_{0} for $|z|\!\leqslant\!1$ m and $\rho_{0}\!=\!0$ elsewhere
- з. $\rho_0 = 20 \epsilon_0$ in the entire region
- $\rho_0 = 20 \epsilon_0$ for $|z| \le 1$ m and $\rho_0 = 0$ elsewhere

Q27 The ratio of work done by an ideal monoatomic gas to the heat supplied to it in an isobaric process is:

Chosen Option : 4

Optio 2

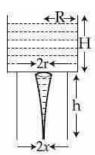
- $2.\frac{3}{2}$
- 3. $\frac{2}{5}$
- 4. $\frac{3}{5}$

Q.28

Chosen Option : --

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Consider a water jar of radius R that has water filled up to height H and is kept on a stand of height h (see figure). Through a hole of radius r ($r \le R$) at its bottom, the water leaks out and the stream of water coming down towards the ground has a shape like a funnel as shown in the figure. If the radius of the cross-section of water stream when it hits the ground is x. Then:

$$x = r \left(\frac{H}{H + h} \right)^{\frac{1}{2}}$$

$$x = r \left(\frac{H}{H+h}\right)^{\frac{1}{4}}$$

$$x = r \left(\frac{H}{H + h} \right)$$

$$4 \quad x = r \left(\frac{H}{H + h}\right)^2$$

Microwave oven acts on the principle of:

Chosen Option: 4

- ns giving translational energy to water molecules
 - 2 giving rotational energy to water molecules
 - transferring electrons from lower to
 - 3 higher energy levels in water molecule
 - giving vibrational energy to water molecules

Q.30

Chosen Option: 1

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When photons of wavelength λ_1 are incident on an isolated sphere, the corresponding stopping potential is found to be V. When photons of wavelength λ_2 are used, the corresponding stopping potential was thrice that of the above value. If light of wavelength λ_3 is used then find the stopping potential for this case:

Options

$$\int_{0}^{8} \frac{hc}{e} \left[\frac{1}{\lambda_3} + \frac{1}{2\lambda_2} - \frac{3}{2\lambda_1} \right]$$

$$2. \frac{hc}{e} \left[\frac{1}{\lambda_3} + \frac{1}{2\lambda_2} - \frac{1}{\lambda_1} \right]$$

3.
$$\frac{hc}{e} \left[\frac{1}{\lambda_3} - \frac{1}{\lambda_2} - \frac{1}{\lambda_1} \right]$$

$$4 \frac{hc}{e} \left[\frac{1}{\lambda_3} + \frac{1}{\lambda_2} - \frac{1}{\lambda_1} \right]$$

Section : Chemistry

Q.1 The hydrocarbon with seven carbon atoms containing a neopentyl and a vinyl group is:

Chosen Option : 1

Optio 1 2, 2-dimethyl-3-pentene

- 2 2, 2-dimethyl-4-pentene
- 3. Isopropyl-2-butene
- 4. 4, 4-dimethylpentene

Q.2 An organic compound contains C, H and S. The minimum molecular weight of the compound containing 8% sulphur is: (atomic weight of S = 32 amu)

Chosen Option : --

Options 1 200 g mol - 1

- 2 300 g mol -1
- 3 400 g mol-1
- 4 600 g mol -1

Q.3 Which one of the following species is stable in aqueous solution?

Chosen Option: 3

optio ns 1 MnO₄³⁻



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- 2. Cr2+
- 3 MnO $_{4}^{2-}$
- 4. Cu +
- Q.4 For the reaction,

 $A(g) + B(g) \rightarrow C(g) + D(g)$, ΔH° and ΔS° are, respectively, -29.8 kJ mol⁻¹ and -0.100 kJ K⁻¹ mol⁻¹ at 298 K. The equilibrium constant for the reaction at 298 K is:

Optio 1 10

- 2 1.0×10^{-10}
- 3. 1.0×10^{10}
- 4. 1
- Q.5 Consider the following sequence for aspartic acid:

Chosen Option : 2

Chosen Option: 4

The pl (isoelectric point) of aspartic acid is:

Optio 1 5.74

- 2.1.88
- 3. 2.77
- 4. 3.65
- Q.6 What will occur if a block of copper metal is dropped into a beaker containing a solution of 1M ZnSO₄?

Chosen Option : 1

- Optio ns 1 The copper metal will dissolve with evolution of hydrogen gas.
 - No reaction will occur.
 - The copper metal will dissolve with evolution of oxygen gas.



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4/1 _C...

W-1-1	l will dissolve and	NonSecured//CBSE165/CBSE165D95/14602995359428164/80105
Q.7 BOD stands for :		Chosen Option : 4
Optio 1 Biochemical Oxid	lation Demand	
² Bacterial Oxidation	on Demand	
3. Biochemical Oxy	gen Demand	
4. Biological Oxyge		
Q.8 The group of mole shape is:	ecules having identical	Chosen Option : 2
Options 1 SF ₄ , XeF ₄ , CCl ₄		
² CIF ₃ , XeOF ₂ , Xel	2 +	
17 14 14 14 14 14 14 14 14 14 14 14 14 14	3	
3 BF ₃ , PCl ₃ , XeO ₃		
⁴ PCl ₅ , IF ₅ , XeO ₂ F	2	
5 L of an alkane requires 25 L of oxygen for its complete combustion. If all volumes are measured at constant temperature and pressure, the alkane is:		Chosen Option : 2
Optio 1 Ethane		
² Propane		
3. Butane		
4 Isobutane		
Q.10 Match the items in use listed in Colun	Column I with its main in II:	Chosen Option : 3
Column I	Column II	
(A) Silica gel	(i) Transistor	
(B) Silicon	(ii) Ion-exchanger	
(b) Cincon	TATIFA SALDI CO	
(C) Silicone	(iii) Drying agent	

 3 (A)-(iii), (B)-(i), (C)-(iv), (D)-(ii) 4 (A)-(iv), (B)-(i), (C)-(ii), (D)-(iii)

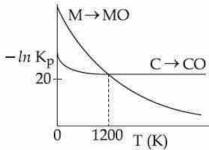
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The plot shows the variation of $-\ln K_p$ versus temperature for the two reactions.

$$M(s) + \frac{1}{2} O_2(g) \rightarrow MO(s)$$
 and

$$\mathsf{C}(\mathsf{s}) + \frac{1}{2}\,\mathsf{O}_2(\mathsf{g}) \,\to\, \mathsf{CO}(\mathsf{s})$$



Identify the correct statement:

Optio At T>1200 K, carbon will reduce MO(s) to M(s).

- 2 At T<1200 K, oxidation of carbon is unfavourable.
- Oxidation of carbon is favourable at all temperatures.

At T<1200 K, the reaction

4 MO(s) + C(s) → M(s) + CO(g) is spontaneous.

Q.12 Bouveault-Blanc reduction reaction involves:

Chosen Option : --

Chosen Option: 1

Reduction of an anhydride with $LiAlH_4$.

- Reduction of a carbonyl compound with Na/Hg and HCl.
- Reduction of an ester with Na/C₂H₅OH.
- Reduction of an acyl halide with H₂/Pd.

Q.13 The gas evolved on heating CH₃MgBr in methanol is:

Chosen Option : 4

Optio 1 Methane

- 2. HBr
- Ethane



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4 Propane	
Q.14 The correct order of the solubility of alkaline-earth metal sulphates in water is:	Chosen Option : 4
Options 1. $Mg > Ca > Sr > Ba$	
2 Mg > Sr > Ca > Ba	
3. Mg < Sr < Ca < Ba	
4 Mg < Ca < Sr < Ba	
Q.15 Identify the correct trend given below: (Atomic No. = Ti: 22, Cr: 24 and Mo: 42)	Chosen Option :
Options Δ_0 of $[Cr(H_2O)_6]^{2+} >$	
1. $[Mo(H_2O)_6]^{2+}$ and	
Δ_0 of $[Ti(H_2O)_6]^{3+} > [Ti(H_2O)_6]^{2+}$	
Δ_0 of $[Cr(H_2O)_6]^{2+}$	
$^{2} < [Mo(H_{2}O)_{6}]^{2+}$ and	
Δ_0 of $[Ti(H_2O)_6]^{3+} > [Ti(H_2O)_6]^{2+}$	
Δ_0 of $[Cr(H_2O)_6]^{2+}$	
$_3 > [Mo(H_2O)_6]^{2+}$ and	
Δ_0 of $[Ti(H_2O)_6]^{3+} \le [Ti(H_2O)_6]^{2+}$	
Δ_0 of $[Cr(H_2O)_6]^{2+}$	
$4 < [Mo(H_2O)_6]^{2+}$ and	
Δ_0 of $[Ti(H_2O)_6]^{3+} < [Ti(H_2O)_6]^{2+}$	
Q.16 At very high pressures, the compressibility factor of one mole of a gas is given by:	Chosen Option : 4
Options 1. $1+rac{pb}{RT}$	
$2. 1 - \frac{b}{(VRT)}$	
3. $\frac{pb}{RT}$	
$4 1 - \frac{pb}{RT}$	
Q.17	Chosen Option : 2



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4/12/2016 https://www.digialm.com///per/g01/pub/223/touchstone/AssessmentQPHTMLNonSecured//CBSE165/CBSE165D95/14602995359428164/80105149_C... The solubility of N2 in water at 300 K and 500 torr partial pressure is 0.01 g L^{-1} . The solubility (in g L -1) at 750 torr partial pressure is: Optio 1 0.02 2. 0.015 3. 0.005 4.0.0075^{Q.18} **Assertion**: Rayon is a semisynthetic Chosen Option : -polymer whose properties are better than natural cotton. Reason: Mechanical and aesthetic properties of cellulose can be improved by acetylation. Both assertion and reason are correct, but the reason is not the correct explanation for the assertion. Both assertion and reason are Assertion is incorrect statement, but the reason is correct. Both assertion and reason are correct. 4 and the reason is the correct explanation for the assertion. Q.19 The artificial sweetener that has the highest Chosen Option: 4 sweetness value in comparison to cane sugar is: Optio 1 Aspartane Saccharin 3. Sucralose 4 Alitame Q.20 Chosen Option: 1



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The reaction of ozone with oxygen atoms in the presence of chlorine atoms can occur by a two step process shown below:

$$O_3(g) + Cl^*(g) \rightarrow O_2(g) + ClO^*(g)$$
 (i)

$$k_i = 5.2 \times 10^9 \, \text{L mol}^{-1} \, \text{s}^{-1}$$

$$CIO^{\bullet}(g) + O^{\bullet}(g) \rightarrow O_2(g) + CI^{\bullet}(g)$$
 (ii)

$$k_{ii} = 2.6 \times 10^{10} L \text{ mol}^{-1} \text{ s}^{-1}$$

The closest rate constant for the overall reaction $O_3(g) + O^{\bullet}(g) \rightarrow 2 O_2(g)$ is:

 $_{\rm ns}^{\rm Optio}$ 1.4×10²⁰ L mol⁻¹ s⁻¹

- 2 3.1 × 10¹⁰ L mol⁻¹ s⁻¹
- $3.5.2 \times 10^9 \, \text{L mol}^{-1} \, \text{s}^{-1}$
- $4.2.6 \times 10^{10} \text{ L mol}^{-1} \text{ s}^{-1}$

Q.21 Identify the incorrect statement regarding heavy water:

Chosen Option: 1

Options It reacts with Al₄C₃ to produce CD₄ and Al(OD)3.

- It reacts with CaC₂ to produce C₂D₂ and Ca(OD)2.
- 3 It is used as a coolant in nuclear reactors.
- It reacts with SO₃ to form deuterated sulphuric acid (D₂SO₄).

Q.22 Which intermolecular force is most responsible in allowing xenon gas to liquefy?

Chosen Option: 4

optio 1. Dipole - dipole

- Ion dipole
- 3. Ionic
- 4 Instantaneous dipole induced dipole

The amount of arsenic pentasulphide that can be obtained when 35.5 g arsenic acid is treated with excess H₂S in the presence of conc. HCl (assuming 100% conversion) is:

Chosen Option : --

Optio 1 0.333 mol



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2 0.125 mol	
3. 0.25 mol	
4 0.50 mol	
Q.24 The non-metal that does not exhibit positive oxidation state is :	Chosen Option : 1
optio 1 Fluorine	
² Chlorine	
3. Iodine	
4 Oxygen	
A particular adsorption process has the following characteristics: (i) It arises due to van der Waals forces and (ii) it is reversible. Identify the correct statement that describes the above adsorption process:	Chosen Option : 3
^{Optio} 1 Adsorption is monolayer.	
2. Enthalpy of adsorption is greater than $100 \text{ kJ} \text{ mol}^{-1}$.	
3 Energy of activation is low.	
Adsorption increases with increase in temperature.	
A reaction at 1 bar is non-spontaneous at low temperature but becomes spontaneous at high temperature. Identify the correct statement about the reaction among the following:	Chosen Option : 4
$_{\rm ns}^{\rm Optio}$ $_{\rm 1}$ ΔH is positive while ΔS is negative.	
² ΔH is negative while ΔS is positive.	
3. Both ΔH and ΔS are negative.	
⁴ Both ΔH and ΔS are positive.	
Q27 The most appropriate method of making egg-albumin sol is:	Chosen Option :
Optio Break an egg carefully and transfer only the yellow part of the content to 100 mL of 5% w/V saline solution and stir well.	



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4/12/2016 https://www.digialm.com///per/g01/pub/223/touchstone/AssessmentQPHTMLNonSecured//CBSE165/CBSE165D95/14602995359428164/80105149_C... Break an egg carefully and transfer the transparent part of the content to 100 mL of 5% w/V saline solution and stir well. Keep the egg in boiling water for 10 minutes. After removing the shell, transfer the white part of the content to 100 mL of 5% w/V saline solution and homogenize with a mechanical shaker. Keep the egg in boiling water for 10 minutes. After removing the shell, transfer the yellow part of the content to 100 mL of 5% w/V saline solution and homogenize with a mechanical shaker. Q.28 Which one of the following complexes will Chosen Option: 1 consume more equivalents of aqueous solution of Ag(NO₃)? Optio 1. [Cr(H₂O)₆]Cl₃ 2. Na₂[CrCl₅(H₂O)] 3. Na₃[CrCl₆] 4 [Cr(H2O)5CI]Cl2 Q.29 The test to distinguish primary, secondary Chosen Option: 1 and tertiary amines is: $^{\mathrm{Optio}}_{\mathrm{ns}}$ $^{\mathrm{1}}$ $\mathrm{C_6H_5SO_2Cl}$ ² Carbylamine reaction Sandmeyer's reaction 4 Mustard oil test The total number of orbitals associated with Chosen Option : 2 the principal quantum number 5 is: Optio 1 25 2. 5 3. 10 4. 20 Section: Mathematics

Chosen Option: 4

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The minimum distance of a point on the curve $y=x^2-4$ from the origin is:

Optio ns 1. $\sqrt{19}$

 $2.\sqrt{\frac{19}{2}}$

 $3.\sqrt{\frac{15}{2}}$

 $\frac{\sqrt{15}}{2}$

Q.2 The shortest distance between the lines

Chosen Option: 1

 $\frac{x}{2} = \frac{y}{2} = \frac{z}{1}$ and $\frac{x+2}{-1} = \frac{y-4}{8} = \frac{z-5}{4}$

lies in the interval:

Optio 1. (2, 3]

2 (3, 4]

3. [0, 1)

4[1, 2)

The value of $\sum_{r=1}^{15} r^2 \left(\frac{15C_r}{15C_{r-1}} \right)$ is equal

Chosen Option : 3

to:

Optio ₁ 560

2.1085

3.680

4 1240

If $\lim_{x \to \infty} \left(1 + \frac{a}{x} - \frac{4}{x^2} \right)^{2x} = e^3$, then 'a' is

Chosen Option : -

equal to:

Optio $\frac{2}{ns}$ $\frac{2}{3}$

 $2.\frac{3}{2}$

3. $\frac{1}{2}$

4. 2



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Q.5 Let x, y, z be positive real numbers	Chosen Option : 1	
such that $x + y + z = 12$ and $x^3y^4z^5 = (0.1) (600)^3$. Then $x^3 + y^3 + z^3$ is		
equal to:		
Optio 1. 216		
2.258		
3. 270		
4 342		
Q.6 A circle passes through (-2,4) and touches	Chosen Option :	
the y -axis at $(0, 2)$. Which one of the	<u> </u>	
following equations can represent a diameter of this circle?		
Option $2x-3y+10=0$		
23x + 4y - 3 = 0		
4x + 5y - 6 = 0		
45x + 2y + 4 = 0		
Q.7		
Let a and b respectively be the semi- transverse and semi-conjugate axes of a	Chosen Option :	
hyperbola whose eccentricity satisfies the		
equation $9e^2 - 18e + 5 = 0$. If S(5, 0) is a focus		
and $5x = 9$ is the corresponding directrix of		
this hyperbola, then $a^2 - b^2$ is equal to:		
Optio 15		
2. 5		
³ -7		
4. 7		
Q.8 16.2 (11.1. (11.1. 2. 1.	Chosen Option : 2	
If $2 \int_0^1 \tan^{-1} x dx = \int_0^1 \cot^{-1} (1 - x + x^2) dx$,		
then $\int_0^1 \tan^{-1}(1-x+x^2) dx$ is equal to :		
Options 1. $\frac{\pi}{2} + \log 2$		
2 log2		
$^{3}\frac{\pi}{2} - \log 4$		
—		
4. log4		
<i>े≅/</i> ठ <i>े</i> न		
Q.9	Chosen Option :	



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If
$$P = \begin{bmatrix} \frac{\sqrt{3}}{2} & \frac{1}{2} \\ -\frac{1}{2} & \frac{\sqrt{3}}{2} \end{bmatrix}$$
, $A = \begin{bmatrix} 1 & 1 \\ 0 & 1 \end{bmatrix}$ and

 $Q = PAP^T$, then $P^T Q^{2015} P$ is:

Optio ns 1.
$$\begin{bmatrix} 0 & 2015 \\ 0 & 0 \end{bmatrix}$$

$$\begin{bmatrix} 2015 & 0 \\ 1 & 2015 \end{bmatrix}$$

If the tangent at a point P, with parameter t, on the curve $x=4t^2+3$, $y=8t^3-1$, $t \in \mathbf{R}$, meets the curve again at a point Q, then the coordinates of Q are :

Chosen Option : --

Option
$$(4t^2+3, -8t^3-1)$$

$$(16t^2+3, -64t^3-1)$$

$$(t^2+3, t^3-1)$$

4
$$(t^2+3, -t^3-1)$$

Q.11 Consider the following two statements :

Chosen Option : 1

- **P:** If 7 is an odd number, then 7 is divisible by 2.
- Q: If 7 is a prime number, then 7 is an odd number.

If V_1 is the truth value of the contrapositive of P and V_2 is the truth value of contrapositive of Q, then the ordered pair (V_1, V_2) equals :

Q.12

Chosen Option : 4



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The point represented by 2+i in the Argand plane moves 1 unit eastwards, then 2 units northwards and finally from there

 $2\sqrt{2}$ units in the south-westwards direction. Then its new position in the Argand plane is at the point represented by:

Options 1. -1-i

- 2. -2 2i
- 3.2 + 2i
- 4.1 + i

Q.13 The point (2, 1) is translated parallel to the

w ne

line L : x-y=4 by $2\sqrt{3}$ units. If the new point Q lies in the third quadrant, then the equation of the line passing through Q and perpendicular to L is :

Optio 1 $x + y = 2 - \sqrt{6}$

- $^{2}x+y=3-2\sqrt{6}$
- $3. x + y = 3 3\sqrt{6}$
- $42x + 2y = 1 \sqrt{6}$

Q.14 If the function

Chosen Option : 3

Chosen Option: --

$$f(x) = \begin{cases} -x, & x < 1 \\ a + \cos^{-1}(x+b), & 1 \le x \le 2 \end{cases}$$

is differentiable at x = 1, then $\frac{a}{b}$ is equal

to

 $\int_{\text{ns}}^{\text{Optio}} \int_{1}^{\pi} \frac{1}{2} = \frac{\pi}{2}$

- $^{2}-1-\cos^{-1}(2)$
- 3. $\frac{\pi + 2}{2}$
- 4. $\frac{\pi 2}{2}$

Q.15

Chosen Option: 3

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Chosen Option: 2

Chosen Option : --

In a triangle ABC, right angled at the vertex A, if the position vectors of A, B and C are

respectively
$$3\hat{i} + \hat{j} - \hat{k}, -\hat{i} + 3\hat{j} + p\hat{k}$$

and $5\hat{i} + q\hat{j} - 4\hat{k}$, then the point (p, q) lies on a line:

Optio 1 parallel to y-axis.

- 2 parallel to x-axis.
- making an obtuse angle with the positive direction of x-axis.
- making an acute angle with the positive direction of *x*-axis.

For $x \in \mathbb{R}$, $x \neq 0$, $x \neq 1$, let $f_0(x) = \frac{1}{1 - x}$ and

 $f_{n+1}(x) = f_0(f_n(x)), n = 0, 1, 2, \dots$ Then the

value of
$$f_{100}(3) + f_1(\frac{2}{3}) + f_2(\frac{3}{2})$$
 is equal

to:

Optio 8

2. $\frac{4}{3}$

3. $\frac{5}{3}$

4. $\frac{1}{3}$

For $x \in \mathbb{R}$, $x \neq -1$, if

$$(1+x)^{2016} + x(1+x)^{2015} + x^2(1+x)^{2014}$$

$$+ \dots + x^{2016} = \sum_{i=0}^{2016} a_i x^i$$
, then a_{17} is equal

to:

Optio 2016!
17! 1999!

2. 2017! 2000!

3. 2016! 16!

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4/ _C...

4 2017! 17! 2000!	
If the tangent at a point on the ellipse $\left[\frac{x^2}{27} + \frac{y^2}{3}\right] = 1$ meets the coordinate axes at A and B, and O is the origin, then the minimum area (in sq. units) of the triangle OAB is: Optio 1. 9 2. $\frac{9}{2}$ 3. $9\sqrt{3}$ 4. $3\sqrt{3}$	Chosen Option :
If the four letter words (need not be meaningful) are to be formed using the letters from the word "MEDITERRANEAN" such that the first letter is R and the fourth letter is E, then the total number of all such words is: Options 1 1! (2!) ³ 2 56 3 110 4 59	Chosen Option :
The area (in sq. units) of the region described by $A = \{(x, y) \mid y \ge x^2 - 5x + 4, x + y \ge 1, y \le 0\}$ is: Options 1 19 6 2 7 2 3 17 6 4 13 6	Chosen Option :
Q.21	Chosen Option : 2



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If A and B are any two events such that $P(A) = \frac{2}{5}$ and $P(A \cap B) = \frac{3}{20}$, then the conditional probability, $P(A|(A' \cup B'))$, where A' denotes the complement of A, is equal to:

- Optio 8/17
 - 2. 5/17
 - 3. $\frac{11}{20}$
 - 4. 1/4

If f(x) is a differentiable function in the interval $(0, \infty)$ such that f(1) = 1 and

Chosen Option : --

 $\lim_{t \to x} \frac{t^2 f(x) - x^2 f(t)}{t - x} = 1, \text{ for each } x > 0,$

then $f(\frac{3}{2})$ is equal to:

Optio 1. 31 18

- $\frac{13}{6}$
- 3. $\frac{25}{9}$
- $4.\frac{23}{18}$

Q.23 If a variable line drawn through the

Chosen Option : --

intersection of the lines $\frac{x}{3} + \frac{y}{4} = 1$ and

 $\frac{x}{4} + \frac{y}{3} = 1$, meets the coordinate axes at

A and B, $(A \neq B)$, then the locus of the midpoint of AB is:

Optio 1 6xy = 7(x+y)

- 2.7xy = 6(x+y)
- $3.14(x+y)^2 97(x+y) + 168 = 0$
- 4 $4(x+y)^2 28(x+y) + 49 = 0$

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The distance of the point $(1, -2, 4)$ from the plane passing through the point $(1, 2, 2)$ and perpendicular to the planes $x-y+2z=3$ and $2x-2y+z+12=0$, is: Options 1 $\sqrt{2}$ 2 $2\sqrt{2}$ 3 $\frac{1}{\sqrt{2}}$ 4 2	Chosen Option : 2
The number of $x \in [0, 2\pi]$ for which $\sqrt{2\sin^4 x + 18\cos^2 x} - \sqrt{2\cos^4 x + 18\sin^2 x}$	Chosen Option :
=1 is: Optio 1. 6 2. 2 3. 8 4. 4	
The number of distinct real roots of the equation, $\begin{vmatrix} \cos x & \sin x & \sin x \\ \sin x & \cos x & \sin x \\ \sin x & \cos x \end{vmatrix} = 0$ in the interval $\left[-\frac{\pi}{4}, \frac{\pi}{4} \right]$ is:	Chosen Option : 3
If m and M are the minimum and the maximum values of $4 + \frac{1}{2}\sin^2 2x - 2\cos^4 x, \ x \in \mathbb{R}, \text{ then}$ $M - m \text{ is equal to :}$	Chosen Option :
Optio ns 1. $\frac{15}{4}$ 2. $\frac{1}{4}$	



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3. $\frac{7}{4}$ 4. $\frac{9}{4}$	
$4\frac{9}{4}$	
Q28 If the mean deviation of the numbers	21
1, 1+d,, 1+100d from their mean is 255,	Chosen Option : 1
then a value of d is:	
2. 10.1	
3. 20.2	
4. 5.05	
If the equations $x^2 + bx - 1 = 0$ and $x^2 + x + b = 0$ have a common root different	Chosen Option : 4
from -1 , then $ b $ is equal to:	
Options 1 $\sqrt{2}$	
2. 2	
3. 3 4. $\sqrt{3}$	
0.30	
If C	Chosen Option :1
$\int \frac{dx}{\cos^3 x \sqrt{2 \sin 2x}} = (\tan x)^A + C(\tan x)^B + k,$	
where k is a constant of integration, then $A + B + C$ equals:	
Options 1. $\frac{16}{5}$ 2. $\frac{7}{10}$ 3. $\frac{21}{5}$ 4. $\frac{27}{10}$	
2. 7 10	
3. <u>21</u> <u>5</u>	
4 27 10	